

REACTIONS OF COORDINATED TRIMETHYLPHOSPHITE WITH BINARY FLUORIDES

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Fluorination of free trimethylphosphite by phosphorus pentafluoride or tungsten hexafluoride involves complex formation followed by rapid F-for-OCH₃ exchange and Michaelis-Arbusov rearrangement reactions (D.W.A. Sharp et al., *J. Chem. Soc. A*, 1969, 872; J.M. Winfield et al., *ibid*, 1970, 501). Reactions between WF₆ or PF₅ and P(OCH₃)₃, coordinated to Fe^{II} (low spin d⁶-inert) or Cu^I (d¹⁰-labile) cations in CH₃CN, counter anions PF₆⁻ or AsF₆⁻, are very different as evidenced by an n.m.r. study.

Reactions between Fe^{II}-P(OCH₃)₃ and WF₆ are very slow at room temperature; the major products are CH₃PF₄ and WOF₄.NCCH₃. Reactions between Cu^I-P(OCH₃)₃ and WF₆ or PF₅ are rapid, even below room temperature, and depend on the stoichiometry. The major products are W₂O₂F₉⁻ and a PF₅X⁻ species, or OPF₃, minor products include CH₃OPF₂, (CH₃O)₂PF, and PF₃. When the mole ratio coordinated P(OCH₃)₃:WF₆ is 1:1, additional W₂O₂F₉ and PF₅X n.m.r. signals are observed.

The reactions involve fluorination of free P(OCH₃)₃ whose concentration in solution is limited by the metal cation, and in the reaction between PF₅ and Cu^I-P(OCH₃)₃ PF₅·P(OCH₃)₃ has been identified as the initial product. Conventional Michaelis-Arbusov rearrangements are of minor importance as CH₃CN acts as a sink for CH₃⁺, but the final step in the formation of CH₃PF₄ is of this type.